

## DITERPENES RELATED TO GRINDELIC ACID AND FURTHER CONSTITUENTS FROM *GRINDELIA* SPECIES\*

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**Key Word Index**—*Grindelia stricta*; *G. paludosa*; *G. camporum*; Compositae; diterpenes; grindelic acid derivatives; tropones; modhephenepoxide.

**Abstract**—The aerial parts of *Grindelia stricta* afforded in addition to known compounds 21 diterpenes, all closely related to grindelic acid. The aerial parts from *G. paludosa* also contain these diterpenes together with modhephenepoxide, while the roots afforded two tropones and a tetralin, which is also present in the leaves. Grindelic acid and several derivatives were also isolated from *G. camporum*. The structures were elucidated by spectroscopic methods. The chemotaxonomic situation is discussed briefly.

### INTRODUCTION

From the large American genus *Grindelia* (tribe Astereae) the occurrence of  $C_{10}$ -acetylenes [1] and several flavones [2–4] has been reported previously. From one species, grindelic acid and its 6-oxo derivative were isolated [5]. We have now investigated *Grindelia stricta*, *G. paludosa* and *G. camporum*. Again, several  $C_{10}$ -acetylenes (1–10), grindelic acid (11a) and 12a were present together with a large number of other diterpenes, all closely related to 11a. Two unusual tropones and modhephenepoxides were also isolated.

### RESULTS AND DISCUSSION

The aerial parts of *G. stricta* afforded the acetylenic esters 1–5, the labdanes grindelic acid (11a) [5], 12a [5], 13a [6], 14a [7] and 36a [8] as well as 21 other diterpenic acids, which could be separated only as their methyl esters.  $^1H$  NMR investigations led to the structures 15b–35b (Tables 1 and 2), consequently 15a–35a were the natural compounds. The structure of 15b was deduced from the corresponding  $^1H$  NMR signals (Table 1) and from decoupling experiments. Irradiation of the broadened doublet at  $\delta$  4.00 collapsed the doublet at 1.68 to a singlet, the broadened singlet at 5.50 to a quartet and the signal of the olefinic methyl to a narrow split doublet, indicating that we were dealing with the signals of H-5, H-6, H-7 and H-17. All other signals were close to those of 11b. The 6a-orientation of the hydroxyl group followed from the coupling  $J_{5,6}$ . The  $^1H$  NMR spectrum of 16b (Table 1) was very similar to that of 15b. However, the H-6 signal was shifted downfield to 4.38 and an additional singlet at 7.45 was present. In the mass spectrum fragments at  $m/z$  348 and 333 ( $C_{21}H_{32}O_4$  and  $C_{21}H_{33}O_3$ ) could be observed, while no molecular ion was present. However, chemical

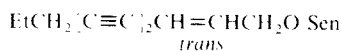
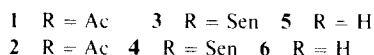
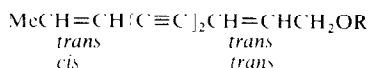
ionization gave a clear  $M + 1$  peak ( $m/z$  379), indicating that  $m/z$  333 probably was formed by loss of  $OCHO$ . Therefore compound 16b was a formate and consequently, mild hydrolysis afforded 15b. The molecular formula of 17b and the IR spectrum of this compound indicated the presence of a further hydroxy derivative of grindelic acid. The  $^1H$  NMR spectrum (Table 1) showed that this OH group could only be placed at C-1 or C-3. Comparison of the chemical shifts of H-18, H-19 and H-20 with those in the spectrum of 11b favoured the C-3 position. The  $^1H$  NMR spectrum of 18b (Table 1) displayed three methyl signals, while the mass spectrum indicated the presence of a norditerpene. The chemical shifts of the methyl signals were in agreement only with the omission of a C-4 methyl group, which was replaced by a hydroxyl. The latter must be  $\alpha$ -orientated, as in the corresponding kaurenes the axial hydroxyl group caused a clear downfield shift of the signal of the C-10 methyl group [9]. Compound 19b was the corresponding formate. All signals (Table 1) were similar to those of 18b, but again an additional downfield singlet indicated the presence of a formate. Only the H-5 double doublet was shifted downfield due to the deshielding effect of the formate carbonyl. The  $^1H$  NMR spectral data of 22b (Table 1) clearly showed that the acetate of 13b was present. In 23b, 24b and 25b the acetate group was replaced by isovalerate. 2-methyl butyrate and isobutyrate, respectively, as could be deduced from the corresponding  $^1H$  NMR signals (Table 1), though these esters could not be separated completely. The  $^1H$  NMR spectrum of 21b (Table 1) showed that an aldehyde was present. The chemical shifts of the methyl signals indicated that this group was  $\alpha$ -orientated at C-4 as the H-20 signal was nearly unchanged. The aldehyde has been prepared previously from 13a [6]. The molecular formula, as well as the  $^1H$  NMR data of 20b, showed that this diterpene was an isomer of 13b. Consequently, the  $^1H$  NMR signals (Table 1) of H-18 and H-19 showed the characteristic differences if compared with those of 13b [10]. Compounds 26b–30b again showed very similar  $^1H$  NMR spectra (Table 2). While 26b–28b could be obtained pure, 29b and 30b still

\* Part 375 in the series "Naturally Occurring Terpene Derivatives". For Part 374, see Bohlmann, F., Adler, A., King, R. M. and Robinson, H. (1982) *Phytochemistry* 21, 173.

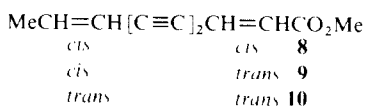
Table 1.  $^1\text{H}$  NMR spectral data of compounds **15b**–**25b** ( $\text{CDCl}_3$ , 400 MHz, TMS as internal standard).

	<b>15b</b>	<b>16b</b>	<b>17b*</b>	<b>18b</b>	<b>19b</b>	<b>20b</b>	<b>21b</b>	<b>22b</b>	<b>23b</b>	<b>24b</b>	<b>25b</b>
H-5	1.68 <i>d</i>	2.08 <i>d</i>			2.28 <i>dd</i>						
H-6	4.00 <i>br d</i>	4.38 <i>br d</i>	1.85 <i>m</i>	1.85 <i>m</i>	1.85 <i>m</i>	1.85 <i>m</i>	1.85 <i>m</i>	1.85 <i>m</i>	1.85 <i>m</i>	1.85 <i>m</i>	1.85 <i>m</i>
H-7	5.50 <i>br s</i>	5.67 <i>br s</i>	5.51 <i>br s</i>	5.53 <i>br d</i>	5.50 <i>br s</i>	5.47 <i>br d</i>	5.42 <i>br s</i>	5.46 <i>br s</i>	5.46 <i>br s</i>	5.46 <i>br s</i>	5.46 <i>br s</i>
H-11	2.05 <i>ddd</i>	2.06 <i>m</i>	2.05 <i>m</i>	2.05 <i>ddd</i>	2.05 <i>ddd</i>	2.08 <i>m</i>	2.08 <i>ddd</i>	2.05 <i>m</i>		2.05 <i>ddd</i>	
H-11'	1.85 <i>m</i>	1.85 <i>m</i>	1.85 <i>m</i>	1.85 <i>m</i>	1.85 <i>m</i>	1.85 <i>m</i>	1.85 <i>m</i>	1.85 <i>m</i>		1.85 <i>m</i>	
H-12	2.20 <i>ddd</i>	2.19 <i>ddd</i>	2.20 <i>ddd</i>	2.20 <i>ddd</i>	2.20 <i>ddd</i>	2.21 <i>ddd</i>	2.20 <i>ddd</i>	2.20 <i>ddd</i>		2.20 <i>ddd</i>	
H-12'	1.88 <i>ddd</i>	1.85 <i>m</i>	1.85 <i>m</i>	1.85 <i>m</i>	1.85 <i>m</i>	1.85 <i>ddd</i>	1.85 <i>ddd</i>	1.85 <i>m</i>		1.85 <i>m</i>	
H-14	2.67 <i>d</i>	2.66 <i>d</i>	2.74 <i>d</i>	2.74 <i>d</i>	2.70 <i>d</i>	2.73 <i>d</i>	2.77 <i>d</i>	2.70 <i>d</i>	2.68 <i>d</i>	2.69 <i>d</i>	2.68 <i>d</i>
H-14'	2.55 <i>br d</i>	2.54 <i>br d</i>	2.60 <i>br d</i>	2.61 <i>br d</i>	2.61 <i>br d</i>	2.59 <i>br d</i>	2.61 <i>br d</i>	2.59 <i>br d</i>	2.58 <i>br d</i>	2.58 <i>br d</i>	2.59 <i>br d</i>
H-16	1.32 <i>s</i>	1.31 <i>s</i>	1.32 <i>s</i>	1.33 <i>s</i>	1.32 <i>s</i>	1.32 <i>s</i>	1.34 <i>s</i>	1.31 <i>s</i>	1.31 <i>s</i>	1.31 <i>s</i>	1.31 <i>s</i>
H-17	1.77 <i>br s</i>	1.83 <i>br s</i>	1.76 <i>br s</i>	1.77 <i>br s</i>	1.77 <i>br s</i>	1.76 <i>br s</i>	1.76 <i>br s</i>	1.76 <i>br s</i>	1.75 <i>br s</i>	1.75 <i>br s</i>	1.76 <i>br s</i>
H-18 } H-18' }	1.12 <i>s</i> }	1.06 <i>s</i> }	0.97 <i>s</i>	-	—	0.88 <i>s</i> }	9.26 <i>s</i>	2.75 <i>d</i> 3.73 <i>d</i>	3.80 <i>d</i> 3.74 <i>d</i>	3.82 <i>d</i> 3.71 <i>d</i>	3.82 <i>d</i> 3.71 <i>d</i>
H-19 } H-19' }	1.00 <i>s</i> }	0.97 <i>s</i> }	0.86 <i>s</i> }	1.18 <i>s</i> }	1.18 <i>s</i>	3.35 <i>d</i> }	1.10 <i>s</i> }	0.91 <i>s</i> }	0.92 <i>s</i> }	0.91 <i>s</i> }	0.92 <i>s</i>
H-20	0.82 <i>s</i>	0.81 <i>s</i>	0.80 <i>s</i>	0.77 <i>s</i>	0.83 <i>s</i>	0.84 <i>s</i>	0.86 <i>s</i>	0.81 <i>s</i>	0.83 <i>s</i>	0.82 <i>s</i>	0.83 <i>s</i>
OCOR		7.45 <i>s</i>	-		7.05 <i>s</i>	-	-	2.04 <i>s</i>	2.20 <i>m</i> 0.94 <i>d</i> 0.93 <i>d</i>	2.37 <i>tq</i> 1.70 <i>ddq</i> 1.48 <i>ddq</i> 0.88 <i>t</i> 1.13 <i>d</i>	2.21 <i>qq</i> 1.16 <i>d</i> 1.15 <i>d</i>
OMe	3.65 <i>s</i>	3.64 <i>s</i>	3.65 <i>s</i>	3.66 <i>s</i>	3.66 <i>s</i>	3.65 <i>s</i>	3.66 <i>s</i>	3.64 <i>s</i>	3.65 <i>s</i>	3.65 <i>s</i>	3.65 <i>s</i>

\* H-3 3.28 *dd* ( $J = 11$  and  $4\text{ Hz}$ );  $J$  (Hz): 11,12 = 9; 11',12 = 8; 11,12' = 5; 11',12' = 8; 12,12' = 13; 14,14' = 14; compounds **15b/16b**: 5,6 = 9.5; 6,7 ~ 2; compound **19b**: 5,6 = 12; 5,6' = 5; compound **20b**: 19,19' = 11; compounds **22b–25b**: 18,18' = 10.5; OiVal: 3',4' = 7; OMeBu: 2',3' = 2',5' = 3',4' = 7; 3',3' = 14; OiBu: 2',3' = 7.



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contained **28b**. The  $^1\text{H}$  NMR spectra, however, clearly showed that we were dealing with esters of grindelic acid where an ester function was introduced at C-17, its nature followed from the typical  $^1\text{H}$  NMR signals. In all cases the signal of the olefinic methyl was replaced by two doublets around  $\delta$  4.5, typical for a  $\text{CH}_2$  group of primary allylic esters. The  $^1\text{H}$  NMR spectrum of **31b** (Table 2) as well as its mass spectrum showed that this compound had an additional carbomethoxy group. The IR data as well as the  $^1\text{H}$  NMR data showed that this ester group was placed equatorially at C-4. Compound **32b** was an oxidation product of **12b**. The  $^1\text{H}$  NMR data (Table 2) indicated that again C-18 was oxygenated bearing an *O*-acetate group. Compound **33b** was an isomer of **14b** as was deduced from the corresponding  $^1\text{H}$  NMR data (Table 2). The  $\alpha$ -orientation of the 7-OH group followed from the hydrogen bond with the ether oxygen (IR:  $3500\text{ cm}^{-1}$ ). Consequently, the OH-proton signal was a clear doublet. Decoupling allowed the assignments of most signals. Compounds **11b–32b** all showed characteristic fragmentations in the mass spectrum. The prominent peak is always

the result of a retro-Diels–Alder fragmentation, which may be followed by elimination of oxygen functions. Compounds **34b** and **35b** were also oxidation products of **11b**. The mass spectrum of **34b** indicated the presence of a bisnorditerpene and the  $^1\text{H}$  NMR data (Table 2) showed that the olefinic part was missing. While many signals still were similar to those of **11b** new double doublets at  $\delta$  2.57 and 2.36, which were coupled with a double doublet at 2.13 indicated the presence of a  $\delta$ -lactone. As no further downfield signals were visible the only possible arrangement of the oxygens was that shown in **34b**. Though the stereochemistry at C-9 could not be determined, the proposed one was most likely from biogenetic considerations. We have named compound **34a** grindelistic acid.

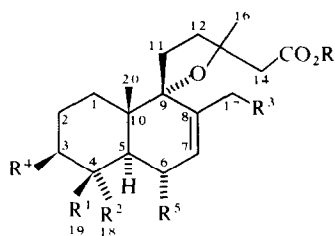
Compound **35b** showed in the  $^1\text{H}$  NMR spectrum (Table 2) the presence of a  $\text{MeCOCH}_2\text{CH}(\text{OR})$  group. As the signal of the proton at the carbon bearing the ether oxygen was coupled with only one other proton and as the remaining signals again were close to those of **11b** the structure and stereochemistry shown for **35b** was most

Table 2.  $^1\text{H}$  NMR spectral data of compounds **26b–35b** ( $\text{CDCl}_3$ , 400 MHz, TMS as internal standard).

	<b>26b</b>	<b>27b</b>	<b>28b</b>	<b>29b</b>	<b>30b</b>	<b>31b</b>	<b>32b</b>	<b>33b*</b>	<b>34b†</b>	<b>35b‡</b>
H-6	1.85 <i>m</i>	1.9 <i>m</i>	1.9 <i>m</i>	1.89 <i>m</i>	1.9 <i>m</i>	—	1.9 <i>ddd</i>	{ 2.57 <i>dd</i> 2.36 <i>dd</i>	4.3 <i>ddd</i>	
H-7	5.93 <i>br s</i>	5.93 <i>br s</i>	5.92 <i>br s</i>	5.12 <i>br s</i>	5.43 <i>br s</i>	5.68 <i>br s</i>	4.26 <i>ddd</i>	—	{ 2.75 <i>dd</i> 2.67 <i>dd</i>	
H-11	1.85 <i>m</i>	1.9 <i>m</i>	1.9 <i>m</i>	1.89 <i>m</i>	2.07 <i>ddd</i>	2.15 <i>m</i>		{ 2.06 <i>m</i> 1.88 <i>m</i>		
H-12 } H-12' }	{ 2.1 <i>m</i> }	{ 2.1 <i>m</i> }	{ 2.1 <i>m</i> }	2.10 <i>m</i>	{ 1.9 <i>m</i> 2.20 <i>ddd</i> 1.90 <i>m</i>	{ 2.0 <i>m</i> 2.3 <i>m</i> 2.0 <i>m</i>	2.2—	{ 2.36 <i>ddd</i> 2.0 <i>m</i> 1.88 <i>m</i>	{ 2.2–1.85 <i>m</i>	
H-14	2.71 <i>d</i>	2.72 <i>d</i>	2.71 <i>d</i>	2.70 <i>d</i>	2.75 <i>d</i>	2.69 <i>d</i>	2.63 <i>d</i>	{ 2.72 <i>s</i> 2.59 <i>br d</i>	2.70 <i>d</i>	
H-14'	2.65 <i>br d</i>	2.65 <i>br d</i>	2.63 <i>br d</i>	2.60 <i>br d</i>	2.65 <i>br d</i>	2.55 <i>br d</i>	2.59 <i>br d</i>		2.65 <i>br d</i>	
H-16	1.33 <i>s</i>	1.35 <i>s</i>	1.34 <i>s</i>	1.34 <i>s</i>	1.33 <i>s</i>	1.40 <i>s</i>	1.31 <i>s</i>	1.30 <i>s</i>	1.20 <i>s</i>	
H-17	4.61 <i>br d</i>	4.63 <i>br d</i>	4.63 <i>br s</i>	4.62 <i>br d</i>	{ 1.75 <i>br s</i> }	{ 1.98 <i>d</i> 4.89 <i>br s</i>	5.09 <i>br s</i>	—	2.22 <i>s</i>	
H-17'	4.53 <i>br d</i>	4.56 <i>br d</i>	4.56 <i>br d</i>	4.50 <i>br d</i>						
H-18 } H-18' }	{ 0.87 <i>s</i> }	{ 0.90 <i>s</i> }	{ 0.89 <i>s</i> }	0.88 <i>s</i>	—	{ 4.30 <i>d</i> 4.10 <i>d</i>	{ 0.91 <i>s</i> }	{ 1.02 <i>s</i> }	0.99 <i>s</i>	
H-19	0.85 <i>s</i>	0.85 <i>s</i>	0.86 <i>s</i>	0.86 <i>s</i>	1.22 <i>s</i>	0.97 <i>s</i>	0.82 <i>s</i>	0.88 <i>s</i>	0.98 <i>s</i>	
H-20	0.77 <i>s</i>	0.80 <i>s</i>	0.79 <i>s</i>	0.79 <i>s</i>	0.82 <i>s</i>	0.86 <i>s</i>	0.74 <i>s</i>	0.86 <i>s</i>	0.92 <i>s</i>	
OCOR	2.06 <i>s</i>	2.35 <i>q</i> 1.14 <i>t</i>	2.20 <i>br d</i> 2.13 <i>m</i> 0.94 <i>d</i>	2.55 <i>tq</i> 1.68 <i>ddq</i> 1.45 <i>ddq</i> 0.92 <i>t</i> 1.13 <i>d</i>	2.36 <i>qq</i> 1.16 <i>d</i> 1.15 <i>d</i>	—	2.03 <i>s</i>	—	—	—
OMe	3.62 <i>s</i>	3.65 <i>s</i>	3.64 <i>s</i>	3.64 <i>s</i>	{ 3.66 <i>s</i> 3.63 <i>s</i>	3.65	3.66 <i>s</i>	3.65 <i>s</i>	3.67 <i>s</i>	

\* OH 4.51 *d*.† H-5 2.13 *dd*.‡ H-5 1.78 *d*.

*J* (Hz): 12,12' = 14,14' = 14; compounds **26b–30b**: 17,17' = 13.5; compound **32b**: 7,17 = 1.5; 18,18' = 10.5; compound **33b**: 5,6 = 12; 6,7 = 2.5; 7, OH = 10; compound **34b**: 5,6 = 6.5; 5,6' = 12.5; 6,6' = 18.5; compound **35b**: 5,6 = 10; 6,7 = 10; 6,7' = 2.5.

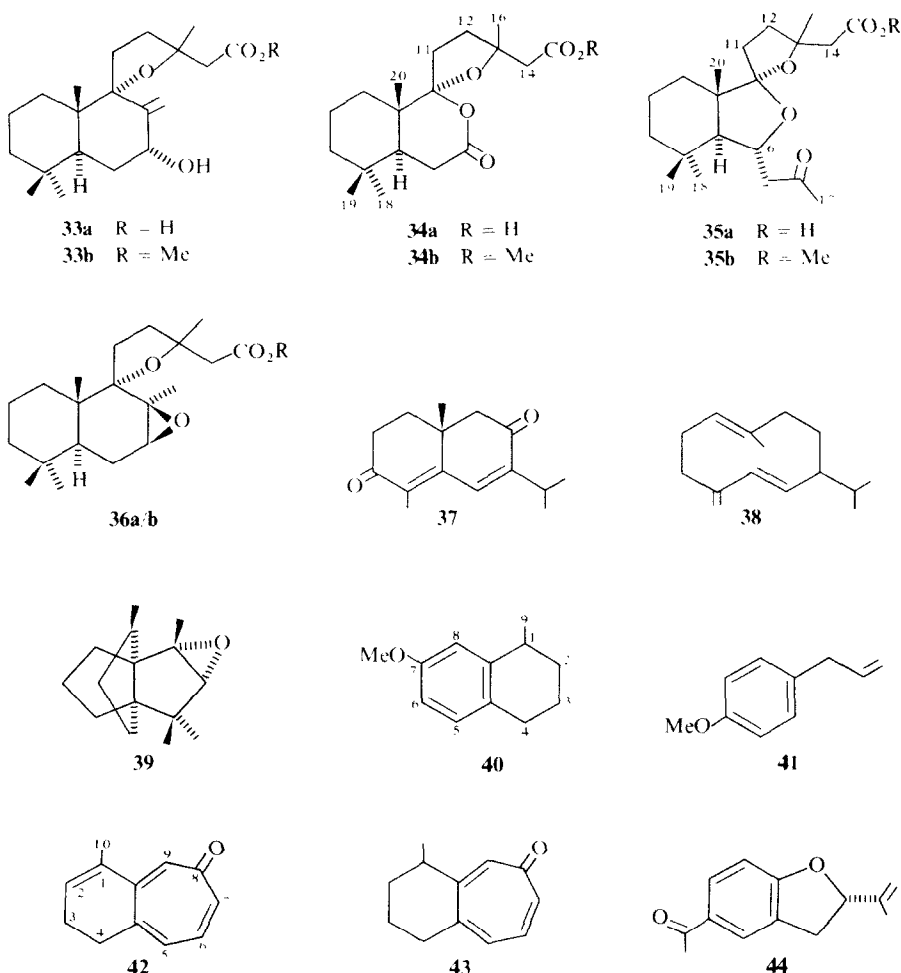


	<b>11a b*</b>	<b>12a b</b>	<b>13a b</b>	<b>14a/b</b>	<b>15a/b</b>	<b>16a/b</b>	<b>17a/b</b>	<b>18a/b</b>	<b>19a/b</b>
R <sup>1</sup>	Me	Me	Me	Me	Me	Me	Me	Me	Me
R <sup>2</sup>	Me	Me	CH <sub>2</sub> OH	Me	Me	Me	Me	OH	OCHO
R <sup>3</sup>	H	H	H	OH	H	H	H	H	H
R <sup>4</sup>	H	H	H	H	H	H	OH	H	H
R <sup>5</sup>	H	=O	H	H	OH	OCHO	H	H	H

	<b>20a/b</b>	<b>21a/b</b>	<b>22a/b</b>	<b>23a/b</b>	<b>24a/b</b>	<b>25a/b</b>
R <sup>1</sup>	CH <sub>2</sub> OH	Me	Me	Me	Me	Me
R <sup>2</sup>	Me	CHO	CH <sub>2</sub> OAc	CH <sub>2</sub> OiVal	CH <sub>2</sub> OMeBu	CH <sub>2</sub> OiBu
R <sup>3</sup>	H	H	H	H	H	H
R <sup>4</sup>	H	H	H	H	H	H
R <sup>5</sup>	H	H	H	H	H	H

	<b>26a/b</b>	<b>27a/b</b>	<b>28a/b</b>	<b>29a/b</b>	<b>30a/b</b>	<b>31a/b</b>
R <sup>1</sup>	Me	Me	Me	Me	Me	Me
R <sup>2</sup>	Me	Me	Me	Me	Me	CO <sub>2</sub> R
R <sup>3</sup>	OAc	OProp	OiVal	OMeBu	OiBu	H
R <sup>4</sup>	H	H	H	H	H	H
R <sup>5</sup>	H	H	H	H	H	H

**32a/b** R<sup>1</sup> = Me, R<sup>2</sup> = CH<sub>2</sub>OAc, R<sup>3</sup> = H, R<sup>4</sup> = H, R<sup>5</sup> = O



likely. The  $6\alpha$ -orientation of the acetonide side-chain was supported by the downfield shift of the H-5 signal. Also, the mass spectrum supported the proposed structure. The loss of an acetone unit from  $\text{M}^+$  and  $\text{M} - \text{CH}_2\text{CO}_2\text{Me}$  is particularly noteworthy. **35a** We have named compound **35a** strictanonic acid.

The aerial parts of *Grindelia paludosa* also afforded grindelic acid (**11a**) and several derivatives (**12a**, **13a**, **15a**, **18a**, **20a**, **22a** and **33a**) and furthermore germacrene D, bornyl and chrysanthemyl acetate, esdragol (**41**), the tetralin derivative **40** and the epoxide **39**. The structure of **40** followed from the molecular formula and the  $^1\text{H}$  NMR spectral data. Compound **40** has been prepared previously by cyclization of the corresponding ester [11]. Compound **39** was identical with the epoxide obtained from modhephene. The roots afforded the acetylenic compounds **2**, **5**, **6** and **8–10**, the diketone **37** [12], the acetophenone derivative **44** and the two tropene derivatives **42** and **43**, whose  $^1\text{H}$  NMR spectral data (Table 3) clearly allowed the assignment of the structures. We have named compound **42** palutropone. While the roots of *Grindelia camporum* afforded **11a**, the aerial parts gave germacrene D,  $\beta$ -farnesene,  $\alpha$ -humulene and bisabolene as well as grindelic acid (**11a**) and the derivatives **12a**, **13a**, **15a**, **16a**, **20a**, **22a**, **33a** and **36a**. Again, all diterpenes were isolated as their methyl esters.

The results obtained show again that  $\text{C}_{10}$ -acetylenes, especially those of types 1–6, may be characteristic for the genus. However, grindelic acid and its derivatives are also typical, though these diterpenes have been isolated from a *Chrysothamnus* [6] and a *Haplopappus* [13] species.

#### EXPERIMENTAL

The air-dried plant material, collected in California, was extracted with  $\text{Et}_2\text{O}$ -petrol (1:2) and the resulting extracts were separated first by CC (Si gel) and further by repeated TLC (Si gel). Known compounds were identified by comparing the IR and  $^1\text{H}$  NMR spectra with those of authentic material.

*Grindelia stricta* DC (voucher RMK 8413, deposited in the U.S. National Herbarium). The aerial parts (320 g) afforded 20 mg **1**, 10 mg **2**, 12 mg **3**, 5 mg **4**, 20 mg **7** and with  $\text{Et}_2\text{O}$ -MeOH (20:1) a complex mixture of acids, which were transformed to the methyl esters by addition of  $\text{CH}_2\text{N}_2$ . TLC using first  $\text{Et}_2\text{O}$  petrol (1:1) and then for the less polar fractions  $\text{CH}_2\text{Cl}_2$ - $\text{C}_6\text{H}_6$  (1:1) and  $\text{CH}_2\text{Cl}_2$ - $\text{C}_6\text{H}_6$ - $\text{Et}_2\text{O}$  (5:5:1 and 2:2:1) for the more polar ones finally gave 30 mg **11b**, 10 mg **12b**, 10 mg **13b**, 50 mg **14b**, 50 mg **15b**, 40 mg **16b**, 15 mg **17b**, 20 mg **18b**, 10 mg **19b**, 10 mg **20b** (separated from **33b** by HPLC, reversed phase; MeOH- $\text{H}_2\text{O}$ , 3:2) 30 mg **21b**, 5 mg **22b**, 10 mg **23b**, 40 mg **24b**, 10 mg **25b**, 5 mg **26b**, 20 mg **27b**, 15 mg **28b**, 10 mg **29b**, 20 mg **30b**, 50 mg **31b**, 10 mg **32b**, 10 mg **33b**, 20 mg **34b**, 20 mg **35b**, and 50 mg **36b**.

Table 3.  $^1\text{H}$  NMR spectral data of compounds **42** and **43** ( $\text{CDCl}_3$ , 400 MHz, TMS as internal standard)

	42	43
H-1	—	2.75 m
H-2	6.29 dd	1.9 m
H-3	2.3 m	
H-4	2.63 br t	2.75 m 2.60 m
H-5	6.81 dd	6.84 br d
H-6	6.98 dd	7.02 dd
H-7	6.87 ddd	6.94 br dd
H-9	7.04 br d	7.06 br d
H-10	2.03 dt	1.28 d

$J$  (Hz): compound **42**: 2,3 = 4.5; 2,10 = 3.10 = 1.5; 3,4 = 7; 5,6 = 8.5; 5,7 = 1; 6,7 = 12; 7,9 = 2.5; compound **43**: 1,10 = 7; 5,6 = 8; 6,7 = 12; 7,9 = 2.5.

*Grindelia paludosa* Greene (voucher RMK 8401, deposited in the U.S. National Herbarium). The roots (80 g) afforded 4 mg **2**, 6 mg **5**, 4 mg **6**, 6 mg **8–10** (ca 2:1:2), 6 mg **37**, 2 mg **40** ( $\text{Et}_2\text{O}$ –petrol, 1:20), 3 mg **42** ( $\text{Et}_2\text{O}$ –petrol, 1:10), 1 mg **43** ( $\text{Et}_2\text{O}$ –petrol, 1:10) and 8 mg **44**, while the aerial parts (900 g) gave 5 mg bornyl acetate, 7 mg chrysanthemyl acetate, 2.5 g **11a**, 100 mg **12a**, 20 mg **13a**, 6 mg **15a**, 5 mg **18a**, 7 mg **20a**, 4 mg **22a**, 10 mg **33a** (all acids isolated as their methyl esters), 4 mg **38**, 1 mg **39** ( $\text{Et}_2\text{O}$ –petrol, 1:20), 4 mg **40** and 3 mg **41**.

*Grindelia camporum* Greene (voucher RMK 8427, deposited in the U.S. National Herbarium). The roots (250 g) afforded 120 mg **11a**, while the aerial parts (650 g) gave 70 mg germacrene D, 40 mg  $\beta$ -farnesene, 70 mg  $\alpha$ -humulene, 80 mg bisabolene and after esterification of the polar fractions 2.7 g **11b**, 30 mg **12b**, 50 mg **13b**, 70 mg **15b**, 60 mg **16b**, 20 mg **20b**, 60 mg **22b**, 30 mg **33b** and 70 mg **36b**.

*Methyl-6 $\alpha$ -hydroxy-grindeloate (15b)*. Colourless oil, IR  $\nu_{\text{max}}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 3600 (OH), 1740 ( $\text{CO}_2\text{R}$ ); MS  $m/z$  (rel. int.): 350.246  $[\text{M}]^+$  (14) ( $\text{C}_{21}\text{H}_{34}\text{O}_4$ ), 332  $[\text{M} - \text{H}_2\text{O}]^+$  (6), 317  $[\text{M} - \text{H}_2\text{O} - \text{Me}]^+$  (2), 226  $[\text{M} - \text{H}_2\text{O} - \text{CHO}]^+$  (60), 135  $[\text{M} - \text{H}_2\text{O} - \text{CH}_2\text{CO}_2\text{Me}]^+$  (62).

$$[\alpha]_{24}^{\text{D}} = \frac{589 \quad 578 \quad 546 \quad 436 \text{ nm}}{-28.2 \quad -28.7 \quad -34.2 \quad -63.6} \quad (c = 0.57, \text{CHCl}_3).$$

*Methyl-6 $\alpha$ -formyloxy-grindeloate (16b)*. Colourless oil, IR  $\nu_{\text{max}}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 1735 ( $\text{CO}_2\text{R}$ ); MS  $m/z$  (rel. int.): 348.230  $[\text{M} - \text{CH}_2\text{O}]^+$  (4) ( $\text{C}_{21}\text{H}_{32}\text{O}_4$ ), 333.243  $[\text{M} - \text{OCHO}]^+$  (4) ( $\text{C}_{21}\text{H}_{33}\text{O}_3$ ), 69  $[\text{C}_5\text{H}_9]^+$  (100); CIMS (iso-butane): 379  $[\text{M} + 1]^+$  (19), 349  $[\text{M} - \text{CH}_2\text{O}]^+$  (67), 333  $[\text{M} - \text{HCO}_2\text{H}]^+$  (24), 145  $[\text{C}_7\text{H}_{13}\text{O}_3]^+$  (100), 127  $[\text{M} - \text{H}_2\text{O}]^+$  (39).

$$[\alpha]_{24}^{\text{D}} = \frac{589 \quad 578 \quad 546 \quad 436 \text{ nm}}{-27.5 \quad -27.5 \quad -32.5 \quad -52.5} \quad (c = 0.4, \text{CHCl}_3).$$

To 4 mg **16b** in 1 ml MeOH 50 mg  $\text{K}_2\text{CO}_3$  in 0.2 ml  $\text{H}_2\text{O}$  was added. Usual work-up after 1 hr afforded **15b**.

*Methyl-3 $\beta$ -hydroxy-grindeloate (17b)*. Colourless oil, IR  $\nu_{\text{max}}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 3640 (OH), 1740 ( $\text{CO}_2\text{R}$ ); MS  $m/z$  (rel. int.): 350.246  $[\text{M}]^+$  (1.6) ( $\text{C}_{21}\text{H}_{34}\text{O}_4$ ), 332  $[\text{M} - \text{H}_2\text{O}]^+$  (2), 301

$[\text{M} - \text{H}_2\text{O} - \text{OMe}]^+$  (2), 210  $[\text{M} - \text{H}_2\text{O} - \text{CHO}]^+$  (100) (RDA).

$$[\alpha]_{24}^{\text{D}} = \frac{589 \quad 578 \quad 546 \quad 436 \text{ nm}}{-23.0 \quad -26.9 \quad -34.6 \quad -61.5} \quad (c = 0.13, \text{CHCl}_3).$$

*Methyl-4 $\alpha$ -hydroxy-18-norgrindeloate (18b)*. Colourless oil, IR  $\nu_{\text{max}}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 3580 (OH), 1740 ( $\text{CO}_2\text{R}$ ); MS  $m/z$  (rel. int.): 336.229  $[\text{M}]^+$  (0.5) ( $\text{C}_{20}\text{H}_{32}\text{O}_4$ ), 318  $[\text{M} - \text{H}_2\text{O}]^+$  (1), 210

$[\text{M} - \text{H}_2\text{O} - \text{OH}]^+$  (100) (RDA).

$$[\alpha]_{24}^{\text{D}} = \frac{589 \quad 578 \quad 546 \quad 436 \text{ nm}}{-32.5 \quad -35.0 \quad -40.5 \quad -70.5} \quad (c = 0.2, \text{CHCl}_3).$$

*Methyl-4 $\alpha$ -formyloxy-18-norgrindeloate (19b)*. Colourless oil, IR  $\nu_{\text{max}}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 1730 ( $\text{CO}_2\text{R}$ ); MS  $m/z$  (rel. int.): 334  $[\text{M} - \text{CH}_2\text{O}]^+$  (5) ( $\text{C}_{20}\text{H}_{30}\text{O}_4$ ), 210 [RDA] (100); CIMS (iso-butane):  $m/z$  365  $[\text{M} + 1]^+$  (7), 335  $[\text{M} - \text{CH}_2\text{O}]^+$  (100), 319  $[\text{M} - \text{HCO}_2\text{H}]^+$  (33).

$$[\alpha]_{24}^{\text{D}} = \frac{589 \quad 578 \quad 546 \quad 436 \text{ nm}}{-35 \quad -38 \quad -45 \quad -75} \quad (c = 0.1, \text{CHCl}_3).$$

*Methyl-19-hydroxy-grindeloate (20b)*. Colourless oil, IR  $\nu_{\text{max}}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 3640 (OH), 1745 ( $\text{CO}_2\text{R}$ ); MS  $m/z$  (rel. int.): 350.246  $[\text{M}]^+$  (1) ( $\text{C}_{21}\text{H}_{34}\text{O}_4$ ), 319  $[\text{M} - \text{CH}_2\text{OH}]^+$  (0.6), 210 [RDA] (100).

$$[\alpha]_{24}^{\text{D}} = \frac{589 \quad 578 \quad 546 \quad 436 \text{ nm}}{-103.5 \quad -107.8 \quad -123.8 \quad -214.5} \quad (c = 0.5, \text{CHCl}_3).$$

*Methyl-18-oxo-grindeloate (21b)*. Colourless oil, IR  $\nu_{\text{max}}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 2710, 1750 (CHO), 1735 ( $\text{CO}_2\text{R}$ ); MS  $m/z$  (rel. int.): 348.230  $[\text{M}]^+$  (0.3) ( $\text{C}_{21}\text{H}_{32}\text{O}_4$ ), 210 [RDA] (100).

$$[\alpha]_{24}^{\text{D}} = \frac{589 \quad 578 \quad 546 \quad 436 \text{ nm}}{-30.9 \quad -32.9 \quad -38.8 \quad -65.9} \quad (c = 0.34, \text{CHCl}_3).$$

*Methyl-18-acetoxygrindeloate (22b)*. Colourless oil, IR  $\nu_{\text{max}}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 1740, 1235 (OAc), 1740 ( $\text{CO}_2\text{R}$ ); MS  $m/z$  (rel. int.): 392.256  $[\text{M}]^+$  (1) ( $\text{C}_{23}\text{H}_{36}\text{O}_5$ ), 332  $[\text{M} - \text{AcOH}]^+$  (3), 210 [RDA] (100).

*Methyl-18-isovaleryloxygrindeloate (23b)*. Colourless oil, not free from **25b**, IR  $\nu_{\text{max}}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 1735 ( $\text{CO}_2\text{R}$ ); MS  $m/z$  (rel. int.): 434.303  $[\text{M}]^+$  (0.5) ( $\text{C}_{26}\text{H}_{42}\text{O}_5$ ), 210 [RDA] (100).

*Methyl-18-[2-methylbutyryloxy]-grindeloate (24b)*. Colourless oil, IR  $\nu_{\text{max}}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 1740 ( $\text{CO}_2\text{R}$ ); MS  $m/z$  (rel. int.): 434.303  $[\text{M}]^+$  (1) ( $\text{C}_{26}\text{H}_{42}\text{O}_5$ ), 332  $[\text{M} - \text{RCO}_2\text{H}]^+$  (1.5), 317  $[\text{M} - \text{Me}]^+$  (1), 210 [RDA] (100).

$$[\alpha]_{24}^{\text{D}} = \frac{589 \quad 578 \quad 546 \quad 436 \text{ nm}}{-15.0 \quad -16.0 \quad -19.5 \quad -38.5} \quad (c = 0.4, \text{CHCl}_3).$$

*Methyl-18-isobutyryloxy-grindeloate (25b)*. Colourless oil, not free from **23b**, IR  $\nu_{\text{max}}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 1735 ( $\text{CO}_2\text{R}$ ); MS  $m/z$  (rel. int.): 420.288  $[\text{M}]^+$  (0.7) ( $\text{C}_{25}\text{H}_{40}\text{O}_5$ ), 210 [RDA] (100).

*Methyl-17-acetoxy-grindeloate (26b)*. Colourless oil, IR  $\nu_{\text{max}}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 1740 ( $\text{CO}_2\text{R}$ ); MS  $m/z$  (rel. int.): 392.256  $[\text{M}]^+$  (2) ( $\text{C}_{23}\text{H}_{36}\text{O}_5$ ), 268 [RDA] (100), 208  $[\text{M} - \text{AcOH}]^+$  (74), 176  $[\text{M} - \text{MeOH}]^+$  (95).

*Methyl-17-propionyloxy-grindeloate (27b)*. Colourless oil, IR  $\nu_{\text{max}}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 1735 ( $\text{CO}_2\text{R}$ ); MS  $m/z$  (rel. int.): 406.272  $[\text{M}]^+$  (0.6) ( $\text{C}_{24}\text{H}_{38}\text{O}_5$ ), 282 [RDA] (38), 208  $[\text{M} - \text{RCO}_2\text{H}]^+$  (48), 134  $[\text{M} - \text{MeCO}_2\text{Me}]^+$  (53), 57  $[\text{Et}_2\text{CO}]^+$  (100).

$$[\alpha]_{24}^{\text{D}} = \frac{589 \quad 578 \quad 546 \quad 436 \text{ nm}}{-21.5 \quad -26.5 \quad -34.0 \quad -58.5} \quad (c = 0.2, \text{CHCl}_3).$$

**Methyl-17-isovaleryloxy-grindeloate (28b).** Colourless oil, IR  $\nu_{\max}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 1735 ( $\text{CO}_2\text{R}$ ); MS  $m/z$  (rel. int.): 434.303  $[\text{M}]^+$  (0.5) ( $\text{C}_{26}\text{H}_{42}\text{O}_5$ ), 310 [RDA] (48), 208  $[\text{310} - \text{RCO}_2\text{H}]^+$  (83), 176  $[\text{208} - \text{MeOH}]^+$  (100), 134  $[\text{208} - \text{MeCO}_2\text{Me}]^+$  (82).

$$[\alpha]_{24}^{25} = \frac{589}{-18.4} - \frac{578}{-19.1} - \frac{546}{-22.0} - \frac{436 \text{ nm}}{-38.3} \quad (c = 1.35, \text{CHCl}_3).$$

**Methyl-17-[2-methylbutyryloxy]-grindeloate (29b).** Colourless oil, not free from **28b**, IR  $\nu_{\max}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 1735 ( $\text{CO}_2\text{R}$ ); MS  $m/z$  (rel. int.): 434.303  $[\text{M}]^+$  (0.7) ( $\text{C}_{26}\text{H}_{42}\text{O}_5$ ), 310 [RDA] (44), 208  $[\text{310} - \text{RCO}_2\text{H}]^+$  (87), 176  $[\text{208} - \text{MeOH}]^+$  (100), 134  $[\text{208} - \text{MeCO}_2\text{Me}]^+$  (72), 85  $[\text{C}_4\text{H}_9\text{CO}]^+$  (20).

**Methyl-17-isobutyryloxy-grindeloate (30b).** Colourless oil, not free from **29b**, IR  $\nu_{\max}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 1735 ( $\text{CO}_2\text{R}$ ); MS  $m/z$  (rel. int.): 420.288  $[\text{M}]^+$  (0.5) ( $\text{C}_{25}\text{H}_{40}\text{O}_5$ ), 296 [RDA] (50), 208  $[\text{296} - \text{RCO}_2\text{H}]^+$  (90), 176  $[\text{208} - \text{MeOH}]^+$  (100).

**Dimethyl-18-oic grindeloate (31b).** Colourless oil, IR  $\nu_{\max}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 1747, 1740 ( $\text{CO}_2\text{R}$ ); MS  $m/z$  (rel. int.): 378.241  $[\text{M}]^+$  (0.5),  $\text{C}_{22}\text{H}_{34}\text{O}_5$ , 319  $[\text{M} - \text{CO}_2\text{Me}]^+$  (0.5), 305  $[\text{M} - \text{CH}_2\text{CO}_2\text{Me}]^+$  (1), 210 [RDA] (100), 136  $[\text{210} - \text{MeCO}_2\text{Me}]^+$  (11).

$$[\alpha]_{24}^{25} = \frac{589}{-15.3} - \frac{578}{-16.1} - \frac{546}{-19.0} - \frac{436 \text{ nm}}{-34.6} \quad (c = 0.52, \text{CHCl}_3).$$

**Methyl-18-acetoxy-6-oxo-grindeloate (32b).** Colourless oil, IR  $\nu_{\max}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 1740 ( $\text{CO}_2\text{R}$ ), 1680 ( $\text{C}=\text{CO}$ ); MS  $m/z$  (rel. int.): 406.236  $[\text{M}]^+$  (2) ( $\text{C}_{23}\text{H}_{34}\text{O}_6$ ), 224 [RDA] (100).

**Methyl-7 $\alpha$ -hydroxy-7,8-dihydro-8(17)-dehydrogrindeloate (33b).** Colourless oil, IR  $\nu_{\max}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 3500 (OH), 1745 ( $\text{CO}_2\text{R}$ ); MS  $m/z$  (rel. int.): 350.246  $[\text{M}]^+$  (10) ( $\text{C}_{21}\text{H}_{34}\text{O}_4$ ), 332  $[\text{M} - \text{H}_2\text{O}]^+$  (12), 236 (100).

$$[\alpha]_{24}^{25} = \frac{589}{-9.5} - \frac{578}{-9.7} - \frac{546}{-11.1} - \frac{436 \text{ nm}}{-15.6} \quad (c = 1.0, \text{CHCl}_3).$$

**Methylgrindelistrictoate (34b).** Colourless oil, IR  $\nu_{\max}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 1750, 1740  $\text{CO}_2$ ; MS  $m/z$  (rel. int.): 294.220  $[\text{M} - \text{CO}_2]^+$  (8) ( $\text{C}_{18}\text{H}_{30}\text{O}_3$ ), 265  $[\text{M} - \text{CH}_2\text{CO}_2\text{Me}]^+$  (8), 173

$[\text{HO}=\text{C}(\text{CH}_2)_4\text{CO}_2\text{Me}]^+$  (62), 109  $[\text{C}_8\text{H}_{13}]^+$  (100); Cl (isobutane): 339  $[\text{M} + 1]^+$  (100).

**Methyl strictanonoate (35b).** Colourless oil, IR  $\nu_{\max}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 1735 ( $\text{CO}_2\text{R}$ ), 1715 ( $\text{C}=\text{O}$ ); MS  $m/z$  (rel. int.): 366.241  $[\text{M}]^+$  (17) ( $\text{C}_{21}\text{H}_{34}\text{O}_5$ ), 335  $[\text{M} - \text{OMe}]^+$  (7), 308  $[\text{M} - \text{Me}_2\text{CO}]^+$  (37), 293  $[\text{308} - \text{Me}]^+$  (25), 280  $[\text{308} - \text{CO}]^+$  (12), 69  $[\text{C}_5\text{H}_9]^+$  (100).

$$[\alpha]_{24}^{25} = \frac{589}{-7.6} - \frac{578}{-7.6} - \frac{546}{-9.5} - \frac{436 \text{ nm}}{-14.2} \quad (c = 0.21, \text{CHCl}_3).$$

**Modhephenepoxide (39).** Colourless oil, MS  $m/z$  (rel. int.): 220.183  $[\text{M}]^+$  (7) ( $\text{C}_{15}\text{H}_{24}\text{O}$ ), 215  $[\text{M} - \text{Me}]^+$  (14), 177  $[\text{215} - \text{CO}]^+$  (18), 55  $[\text{C}_4\text{H}_7]^+$  (100). Modhephene (4 mg) in 1 ml

$\text{CH}_2\text{Cl}_2$  was stirred with 10 mg  $\text{NaHCO}_3$  and 4 mg *m*-chloroperbenzoic acid for 30 min. TLC ( $\text{Et}_2\text{O}$ -petrol, 1:15) afforded 2 mg **39**, identical with the natural compound ( $^1\text{H}$  NMR, TLC).

**1-Methyl-7-methoxytetralin (40).** Colourless oil, IR  $\nu_{\max}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 2880, 1620, 1510, 1390, 1140; MS  $m/z$  (rel. int.): 176.120  $[\text{M}]^+$  (100) ( $\text{C}_{12}\text{H}_{16}\text{O}$ ), 161  $[\text{M} - \text{Me}]^+$  (95).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz): 2.88 (*br ddq*, H-1), 1.88 *m*, 1.50 (*m*, H-2), 1.85 *m*, 1.69 (*m*, H-3), 2.69 (*m*, H-4), 6.99 (*br d*, H-5), 6.67 (*dd*, H-6), 6.71 (*d*, H-8), 1.27 (*d*, H-9), 3.77 (*s*, OMe), [*J* (Hz): 1.2 = 1.9 = 7: 5.6 = 8.5; 6.8 = 2.5].

**Palutropone (42).** Colourless oil, IR  $\nu_{\max}^{\text{CCl}_4}$   $\text{cm}^{-1}$ : 1640, 1580, 910; MS  $m/z$  (rel. int.): 172.089  $[\text{M}]^+$  (8) ( $\text{C}_{12}\text{H}_{12}\text{O}$ ), 144.094  $[\text{M} - \text{CO}]^+$  (38) ( $\text{C}_{11}\text{H}_{12}$ ), 129  $[\text{144} - \text{Me}]^+$  (100).

**Dihydropalutropone (43).** Colourless oil, MS  $m/z$  (rel. int.): 174.104  $[\text{M}]^+$  (17), ( $\text{C}_{12}\text{H}_{14}\text{O}$ ), 146  $[\text{M} - \text{CO}]^+$  (16), 131  $[\text{146} - \text{Me}]^+$  (100).

$$[\alpha]_{24}^{25} = \frac{589}{+18} - \frac{578}{+19} - \frac{546}{+20} - \frac{436 \text{ nm}}{+34} \quad (c = 0.1, \text{CHCl}_3).$$

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